

Chapter 21

Question 21.7

(i) By what percentage would the osmotic concentration of extracellular fluids decrease?

The 50 mL intake of water is entirely absorbed from the gut and rapidly distributes through 3.5 L (3500 mL).

Therefore, the total resulting volume of fluids = $50 + 3500 = 3550$ (mL)

Hence, the proportional dilution of the body fluids by water intake = $3500/3550 = 0.9859$

And, the proportional decrease in osmolality of extracellular fluids (ECF) = $1 - 0.9859 = 0.0141$

So, **percent decrease in osmotic concentration of extracellular fluids = 1.41 %**

(ii) How much (in mOsm kg⁻¹) would plasma osmolality decrease after consuming the water?

Proportional dilution of ECF calculated in answering question (i) = 0.0141

Therefore, decrease in ECF osmolality = $0.014 \times 300 \text{ mOsm kg}^{-1} = \mathbf{4.23 \text{ mOsm kg}^{-1}}$

(iii) The calculated decrease in ECF osmolality is sufficient to reduce secretion of arginine vasopressin, as we learn in Section 21.1.4.

Question 21.12

(i) Rate of Ca²⁺ filtration = GFR × Ca²⁺ concentration in filtrate Equation 1

The glomerular filtration rate (GFR) is given in the question as 0.2 mL min^{-1} .

The Ca²⁺ concentration in the glomerular filtrate is determined by the total Ca²⁺ concentration of the plasma of the mice (given as 2.5 mmol L^{-1}) of which 60 per cent is stated to be filterable.

Therefore:

Ca²⁺ concentration in the glomerular filtrate = $(60/100) \times 2.5 \text{ (mmol L}^{-1}) = 1.5 \text{ mmol L}^{-1} = 1.5 \div 1000 \text{ mmol mL}^{-1}$

Inserting Ca²⁺ and GFR in Equation 1 gives:

Rate of Ca²⁺ filtration (mmol min⁻¹) = $0.2 \text{ (mL min}^{-1}) \times (1.5 \div 1000) \text{ (mmol mL}^{-1})$
 $= 0.0003 \text{ mmol min}^{-1}$

The question asks for calculation of the rate of Ca²⁺ filtration in $\mu\text{mol h}^{-1}$

**Butler, Brown, Stephenson & Speakman, *Animal Physiology*
Solutions to numerical exercises**

To convert rate of Ca^{2+} filtration of $0.0003 \text{ mmol min}^{-1}$ to $\mu\text{mol min}^{-1}$ we need to multiply the value by 1000, and to convert value in $\mu\text{mol min}^{-1}$ to $\mu\text{mol h}^{-1}$ we need to multiply value by 60. Therefore:

$$\begin{aligned}\text{Rate of } \text{Ca}^{2+} \text{ filtration } (\mu\text{mol h}^{-1}) &= 0.0003 \times 1000 \times 60 \\ &= \mathbf{18 \mu\text{mol h}^{-1}}\end{aligned}$$

(ii) The question tells us that the distal nephron reabsorbs 9 per cent of the filtered Ca^{2+}

$$\begin{aligned}9 \text{ per cent of filtered } \text{Ca}^{2+} &= (9/100) \times 18 (\mu\text{mol h}^{-1}) \\ &= \mathbf{1.62 \mu\text{mol h}^{-1}}\end{aligned}$$